

Reduced Nitrogen in the Western US in Winter



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Sensitivity of $\text{PM}_{2.5}$ to NH_3

For dry particles, the presence of excess NH_3 allows semi-volatile salts to form, e.g. NH_4NO_3

For deliquesced particles (and liquid water clouds), NH_3 controls the pH of $\text{PM}_{2.5}$

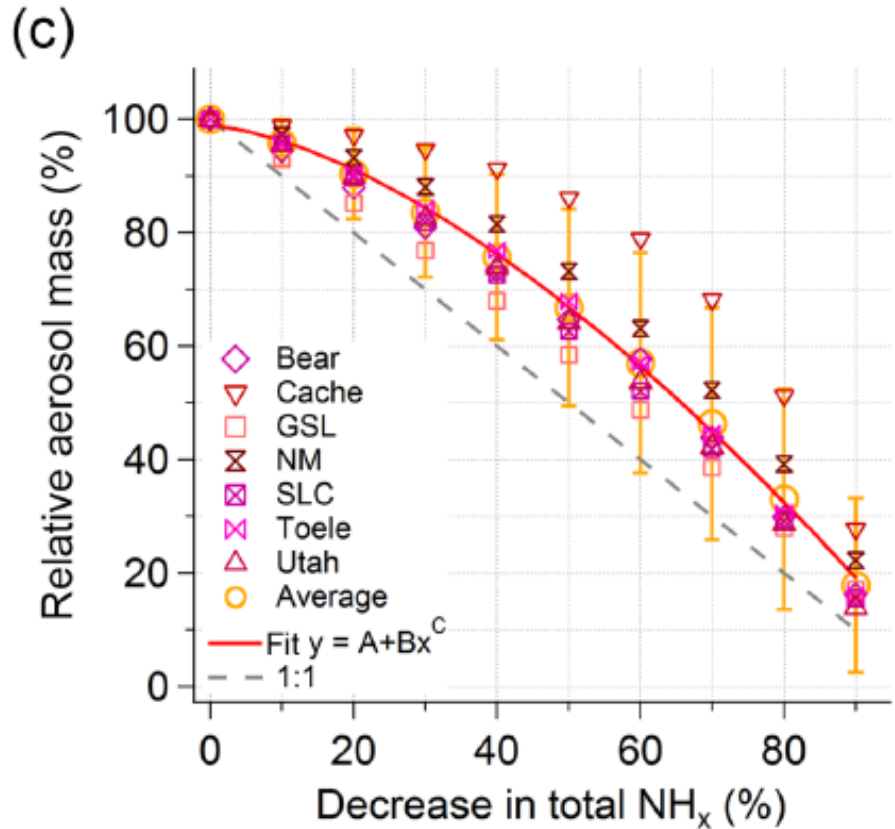
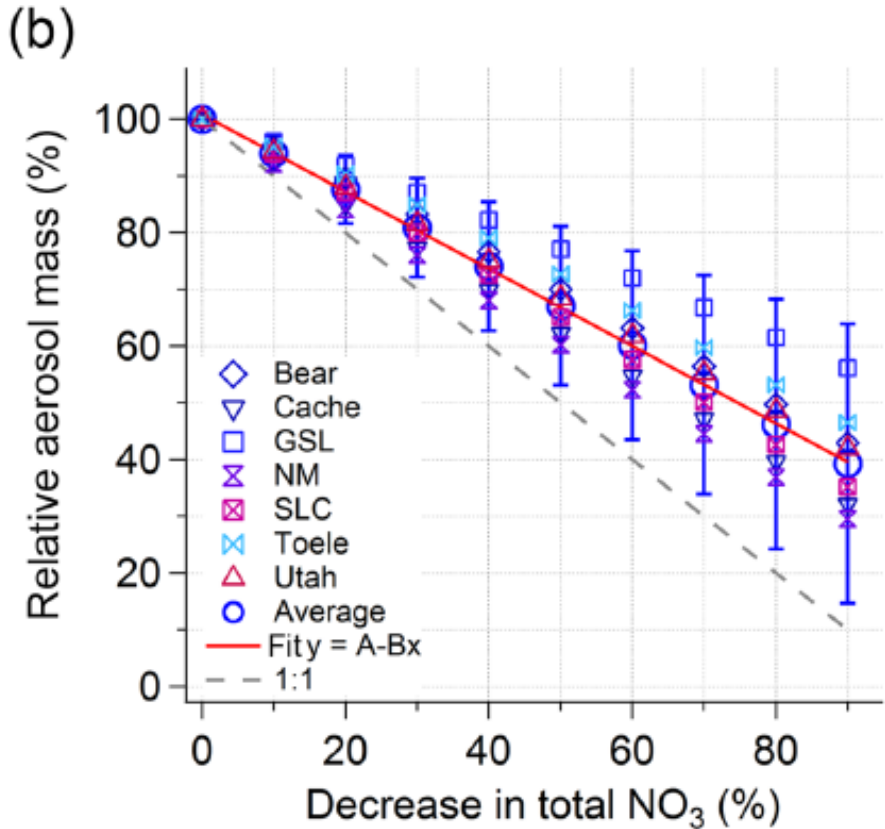
- partitioning of semi-volatile gases, e.g. HNO_3 , HCl , oxalic acid

- rates of some aqueous phase reactions

NH_3 can also react with particle phase carbonyls to generate brown (light-absorbing) constituents

Deliquesced NH_4NO_3 is limited by NH_3 and NO_x

aircraft AMS, CIMS and QCL data from UWFPs 2017 in Utah



Sources of NH₃

Livestock

Fertilizer application

Residential wood combustion

Biomass burning

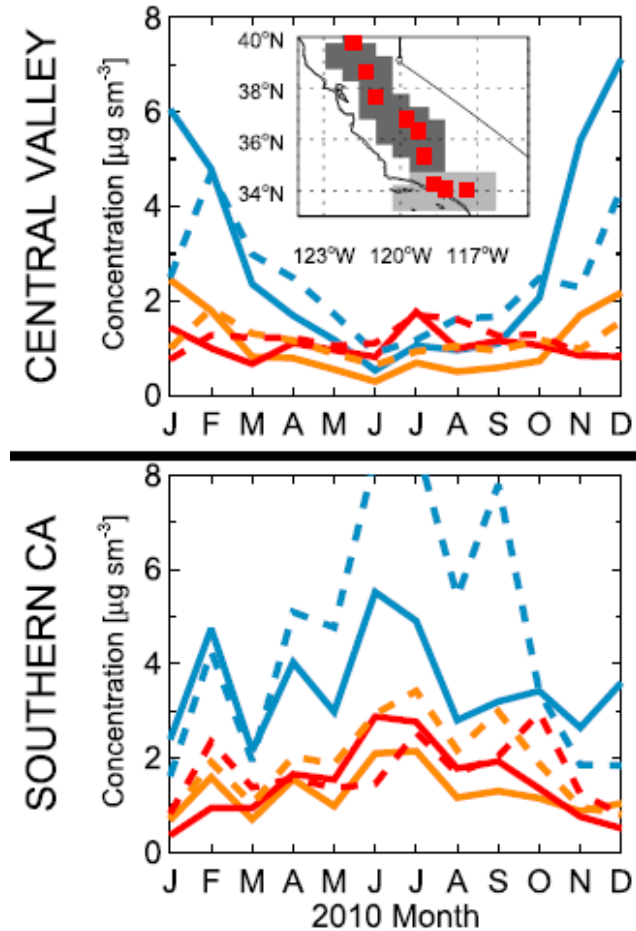
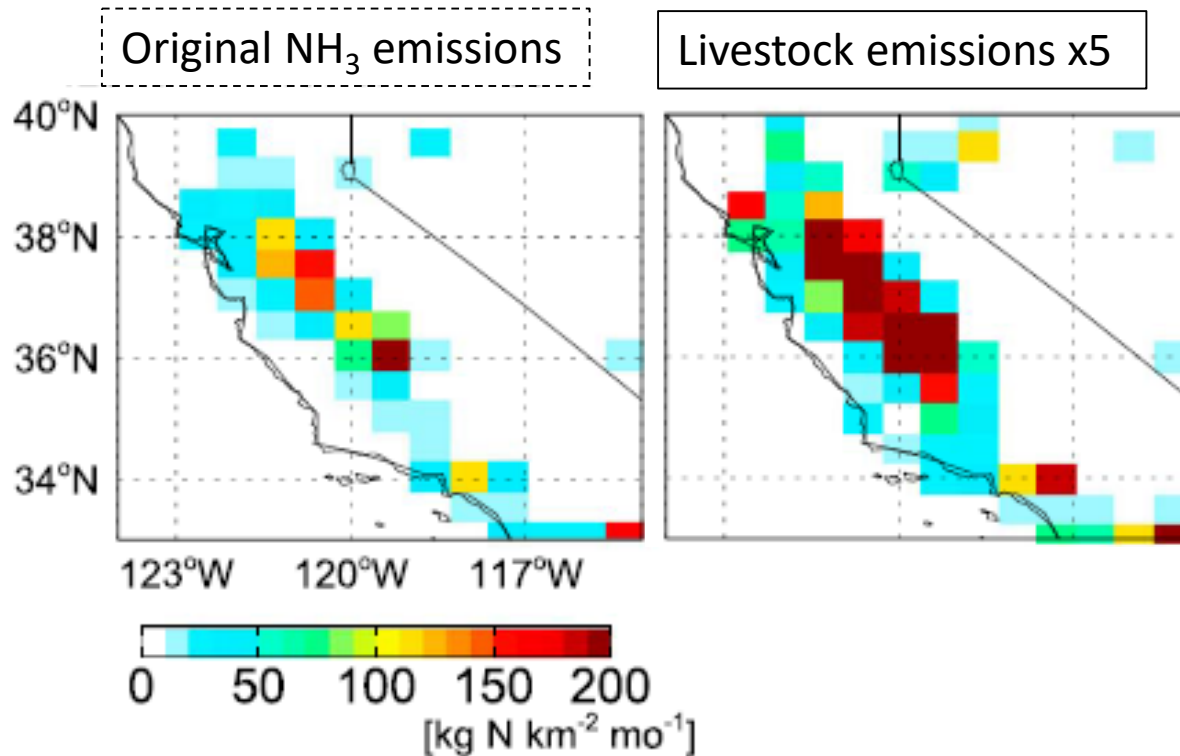
Mobile sources

Industry

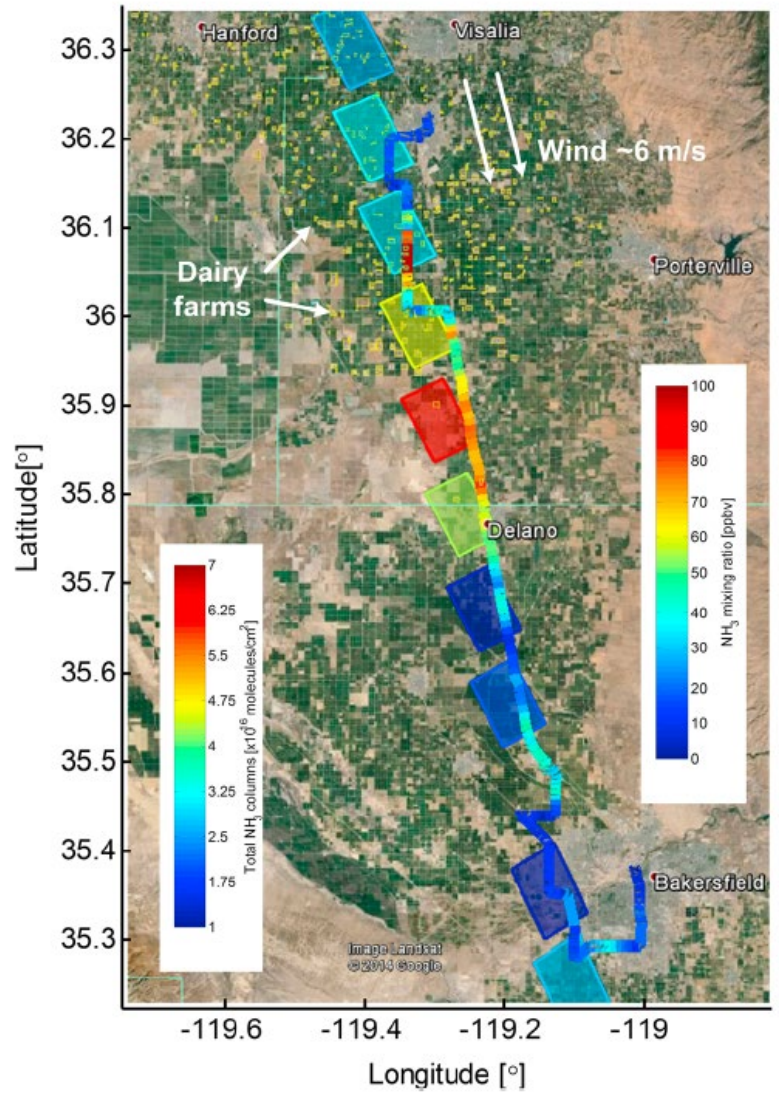
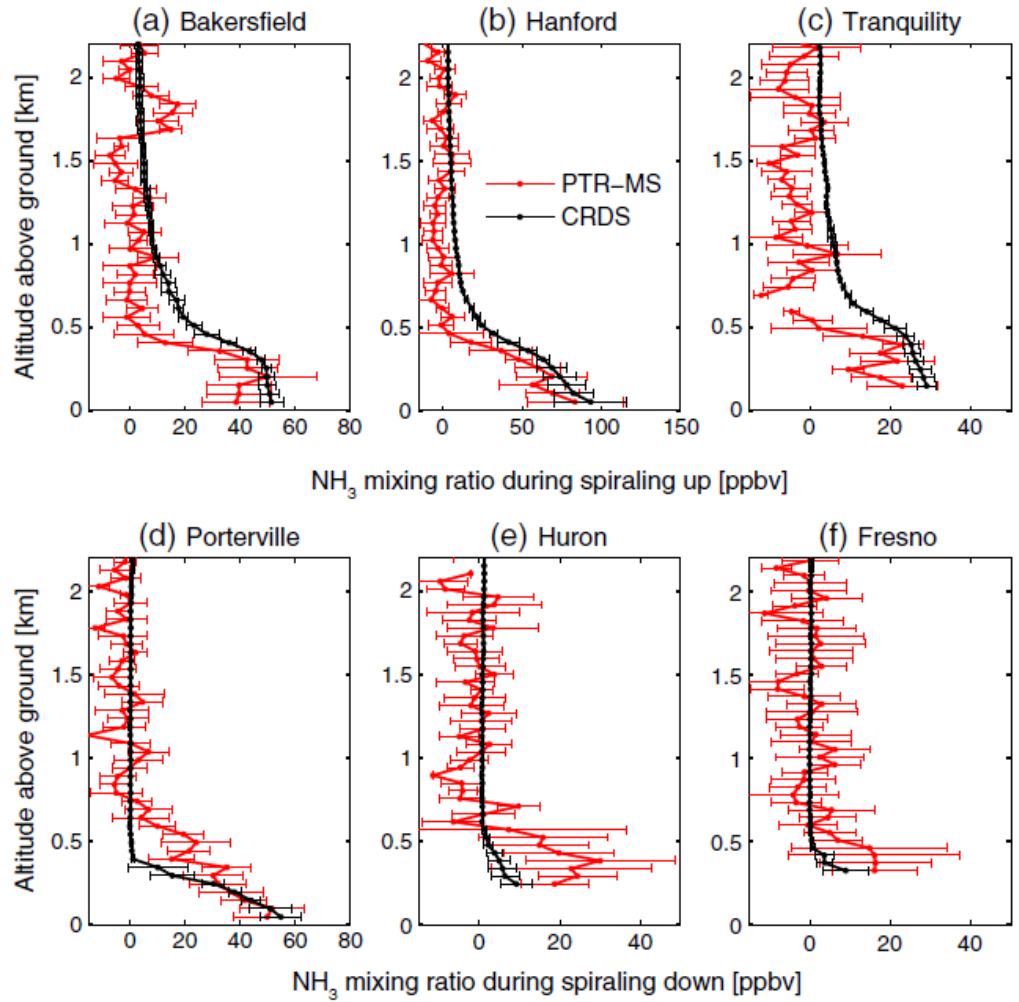
Bidirectional exchange

CalNex implied livestock NH_3 sources were significantly underestimated

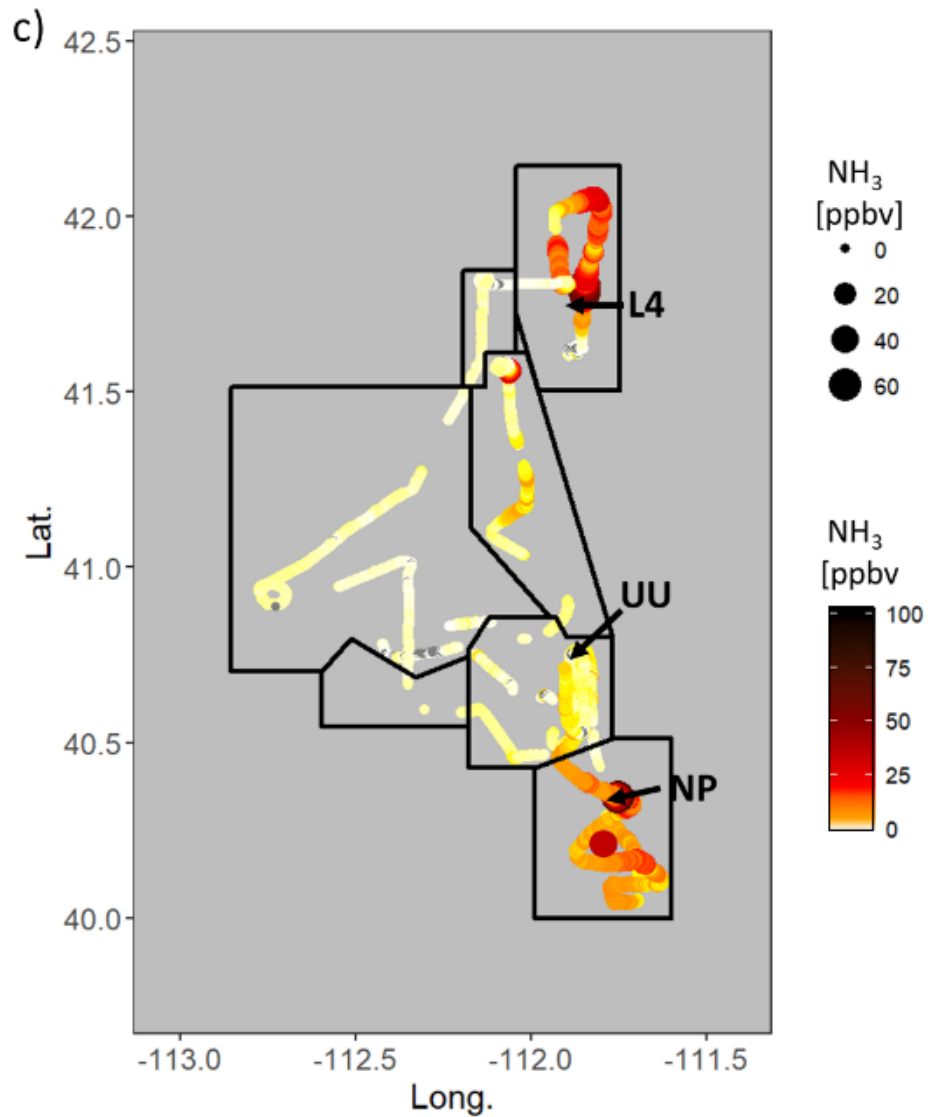
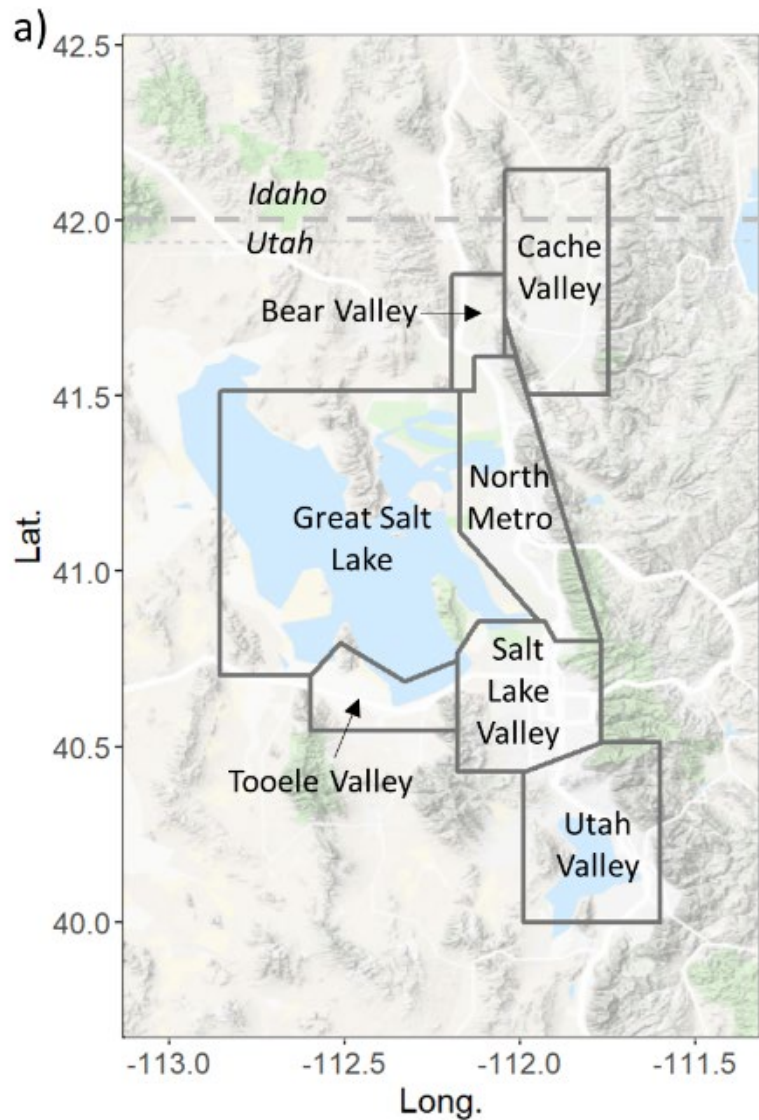
GEOS-Chem modelling using NEI 2005



Satellite, Aircraft and Ground-based in DISCOVER-AQ

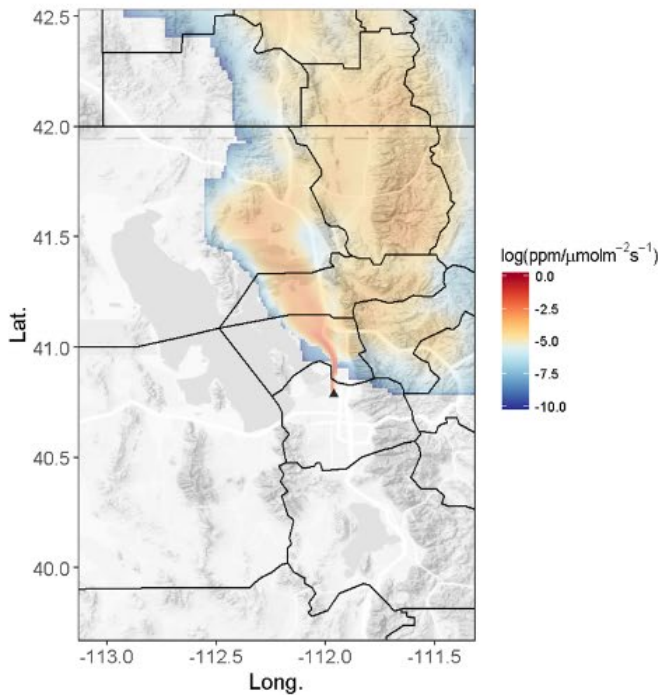


Spatial distribution of NH_3 in UWFPS



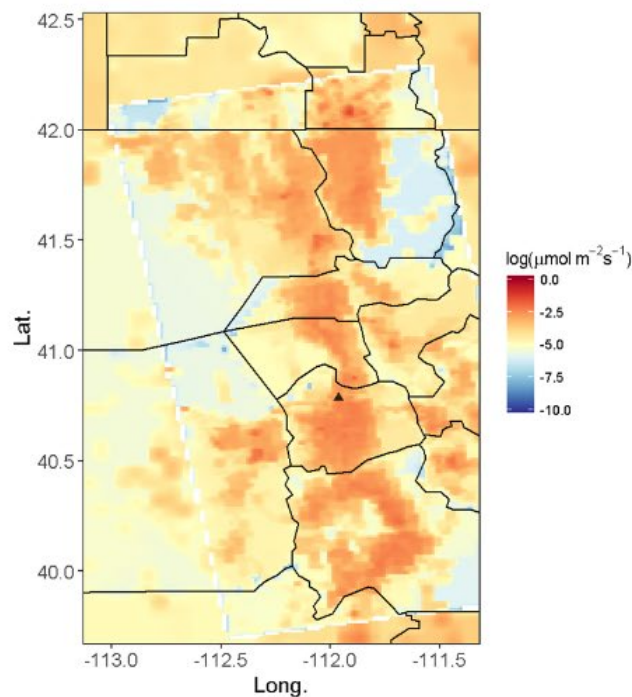
Constraining the NH₃ inventory

STILT footprint



Using HRRR met fields at 3x3 km

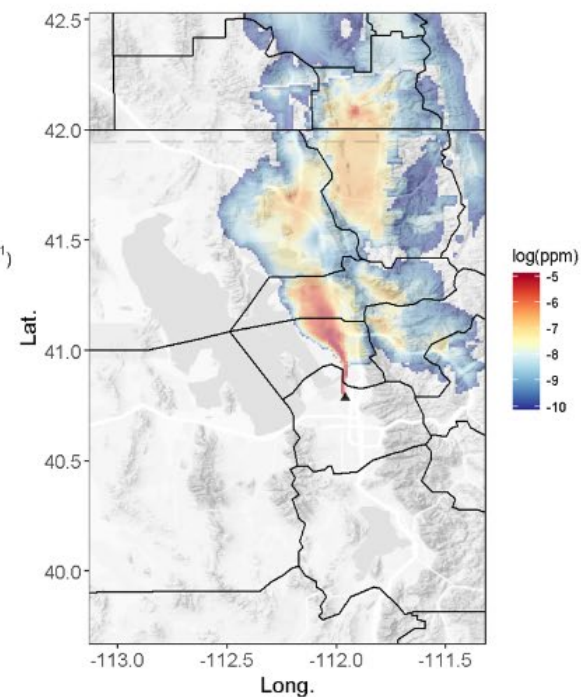
NH₃ emission inventory



UDAQ based on NEI 2014

=

NH₃ contribution

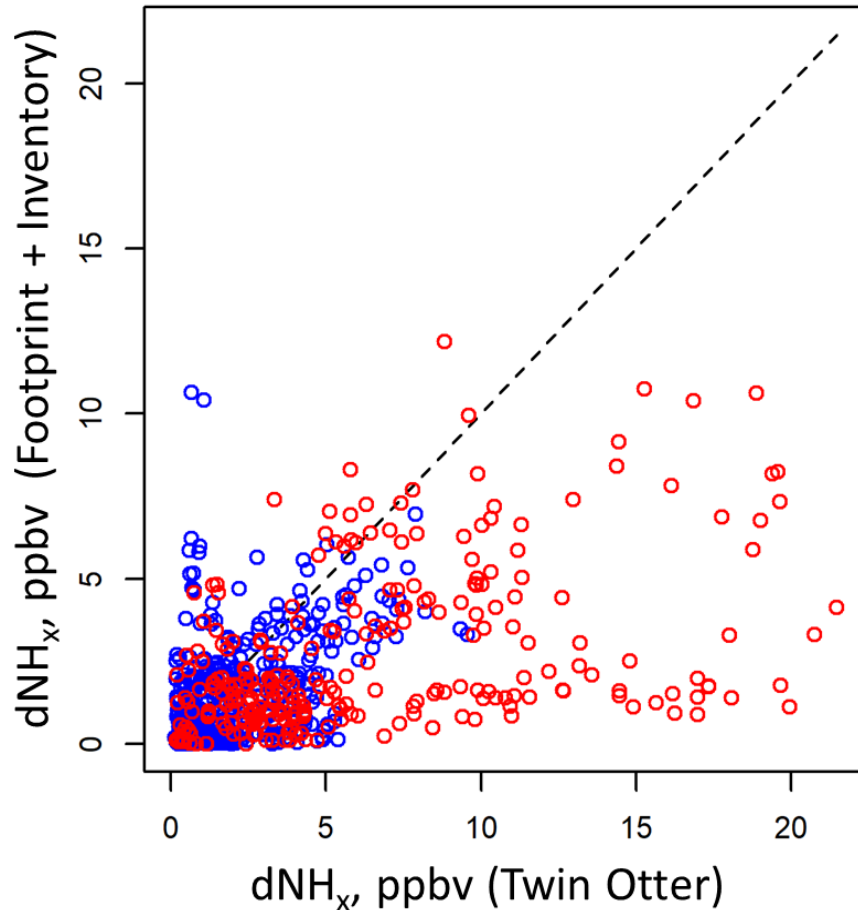


sum
NH₃ mixing ratio at receptor (0.87 ppbv) ⁸

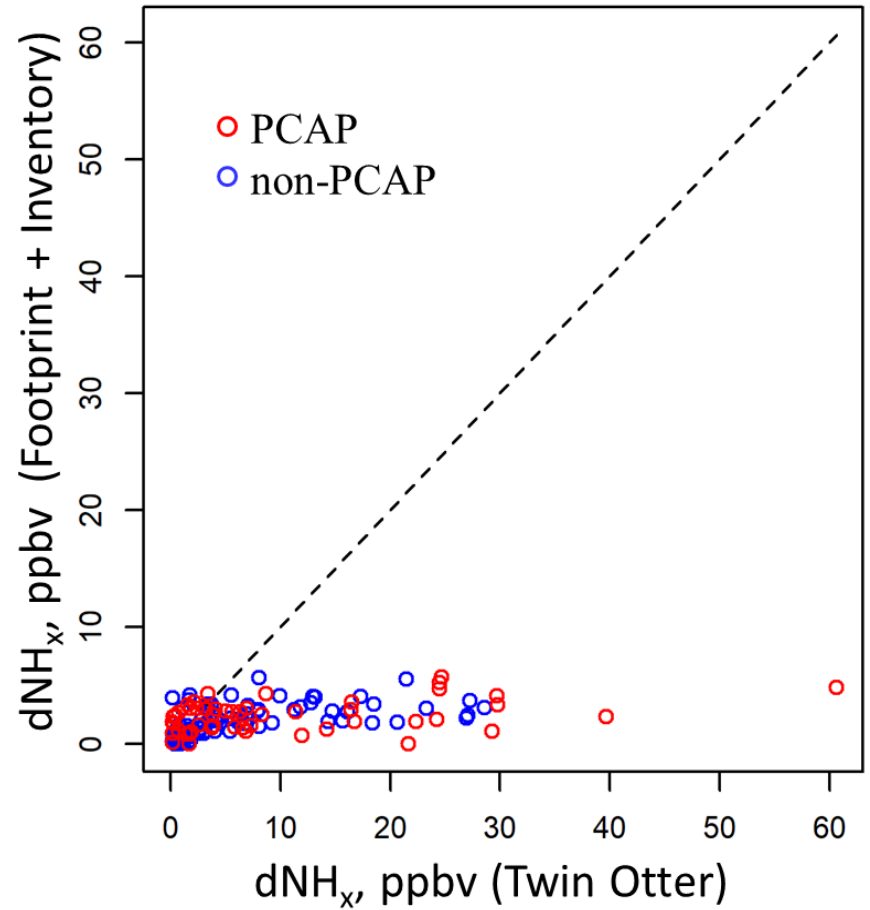
Compare to measurements

Underestimate of NH_3 emissions in Utah

Salt Lake Valley

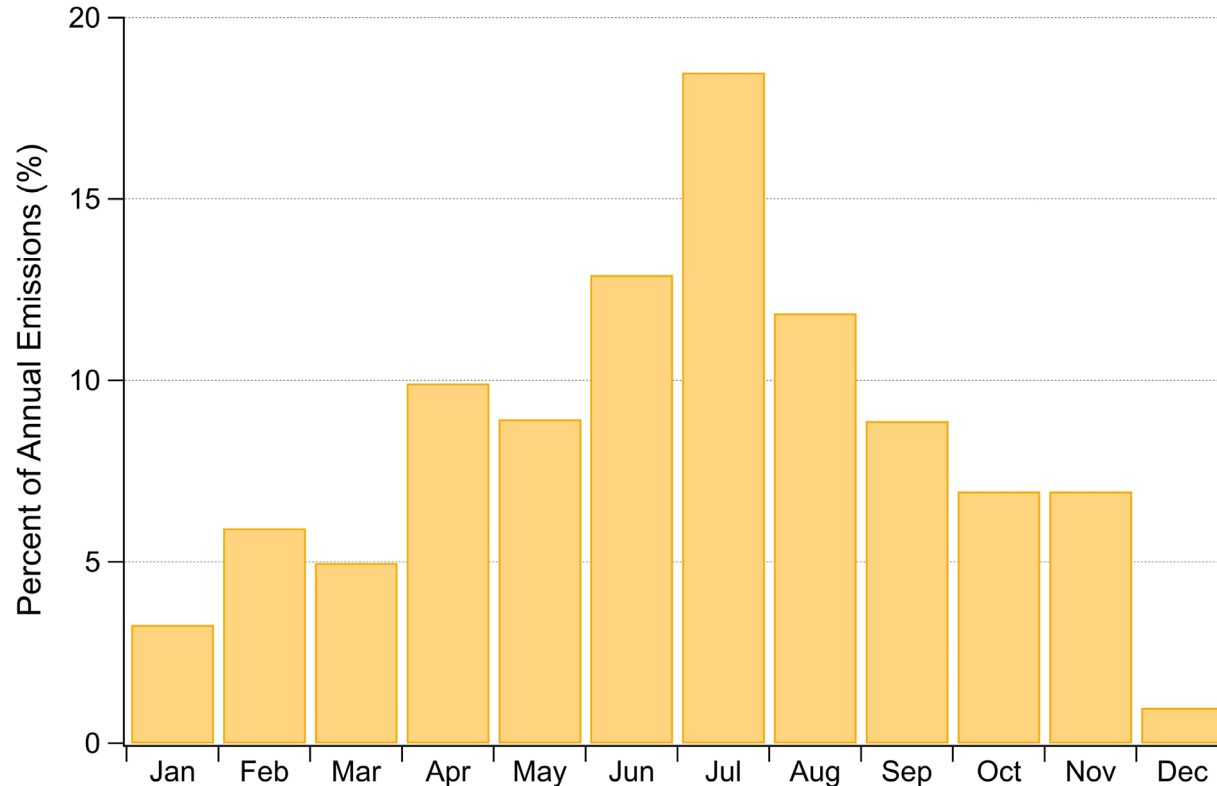


Cache Valley



Wintertime emissions from animal husbandry 4x too low in 2014 UDAQ inventory

Large seasonality for livestock NH_3 emissions in UDAQ inventory



In contrast, USU inventory from Moore and Martin has equivalent emissions in July and January

Seasonal cycle imposed on annual emissions
Cycle was inferred through inverse modelling in Gilliland et al., 2006

CAFO survey in Colorado showed modest seasonality

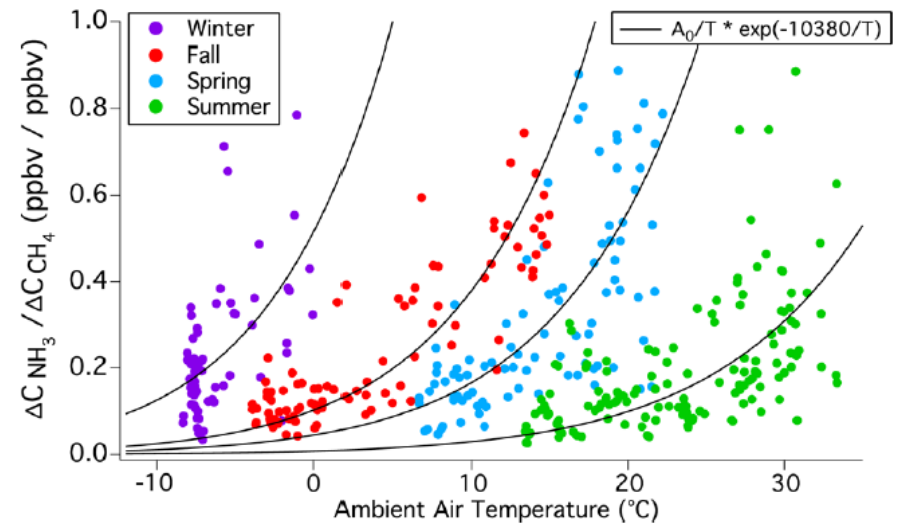
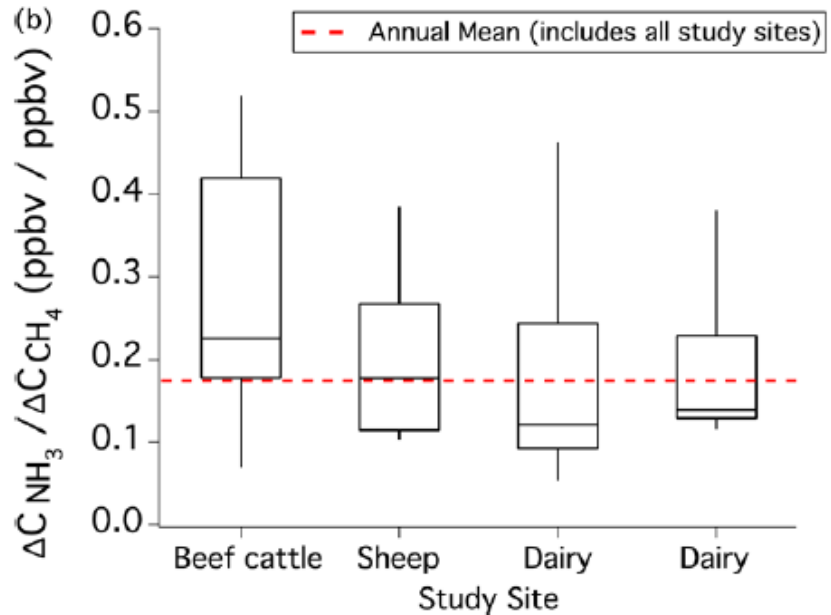
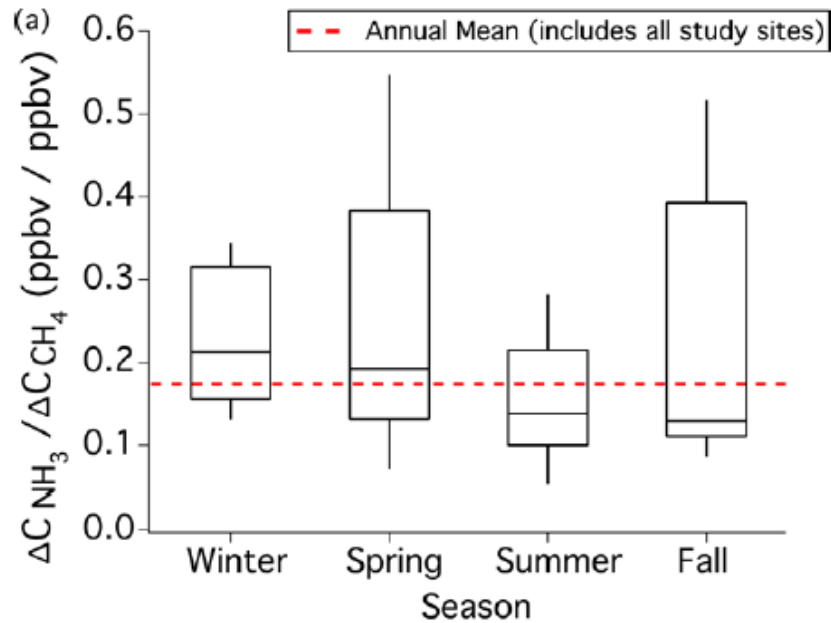
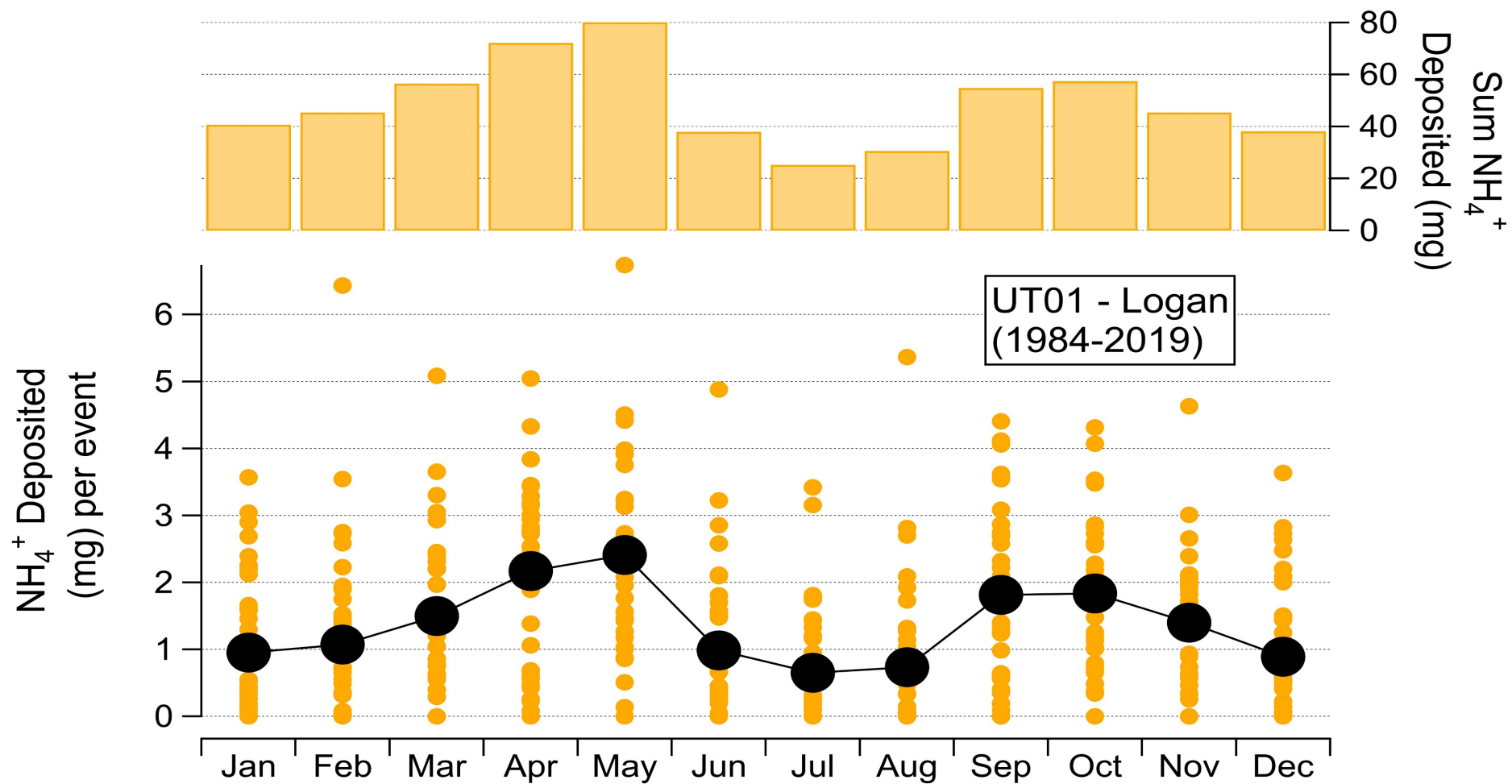
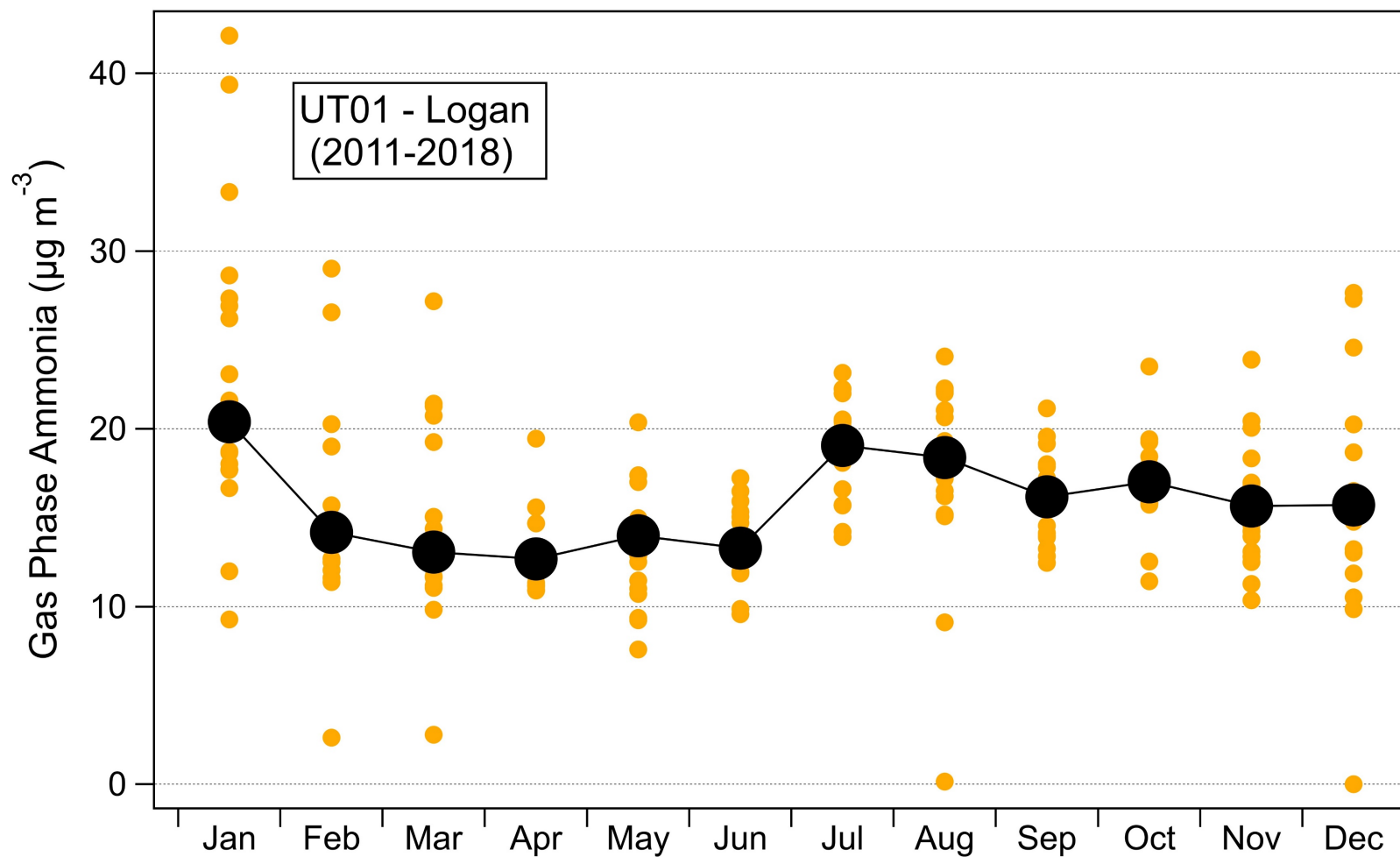


Figure 6. Temperature dependence of the $\Delta C_{\text{NH}_3} / \Delta C_{\text{CH}_4}$ enhancement ratio in each season. The black lines represent the temperature dependence of the ammonia volatilization process and are scaled by a multiplicative factor A_0 for each season.

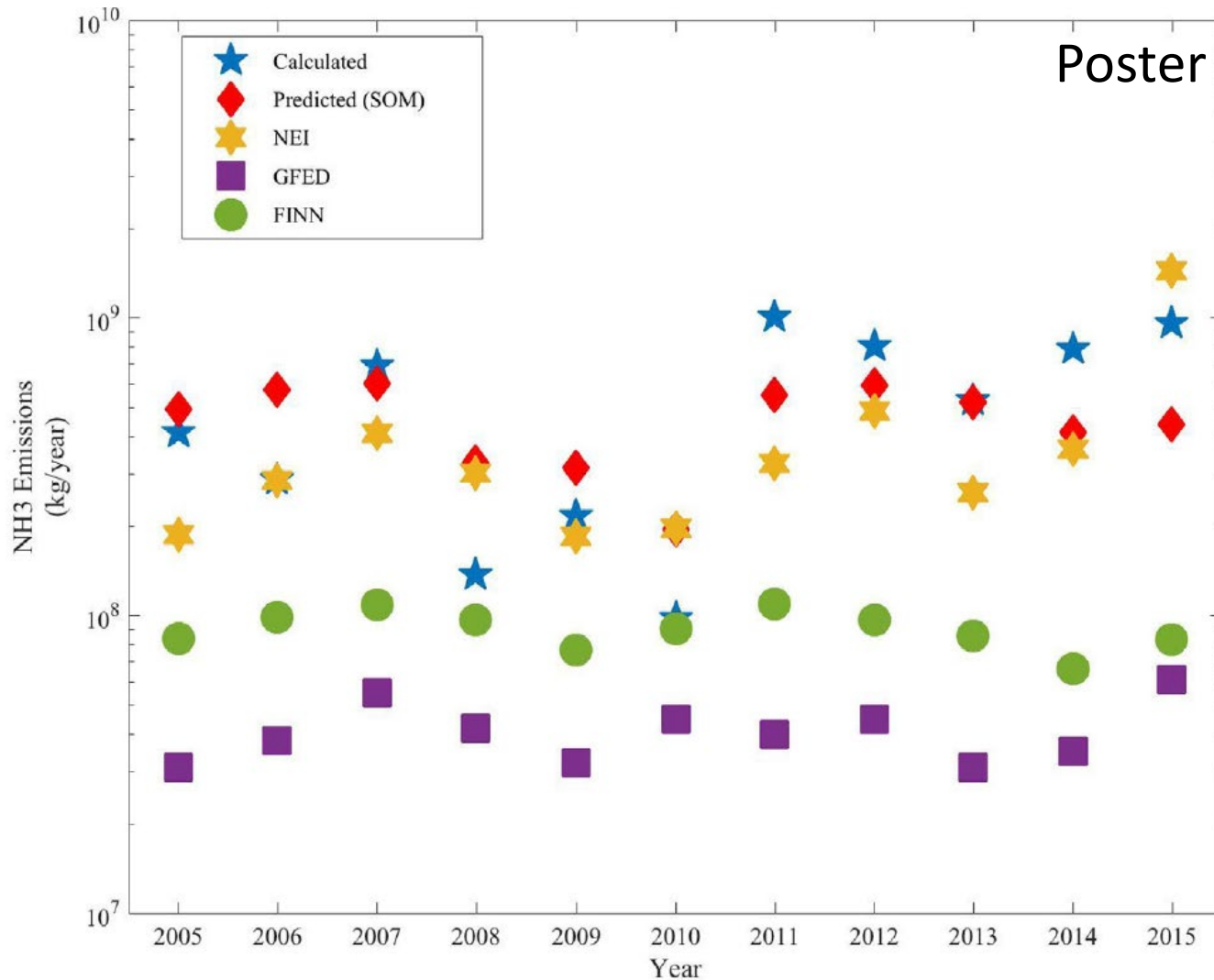
NH_x deposition at Logan (NADP)



NH₃ concentration at Logan (AMoN)

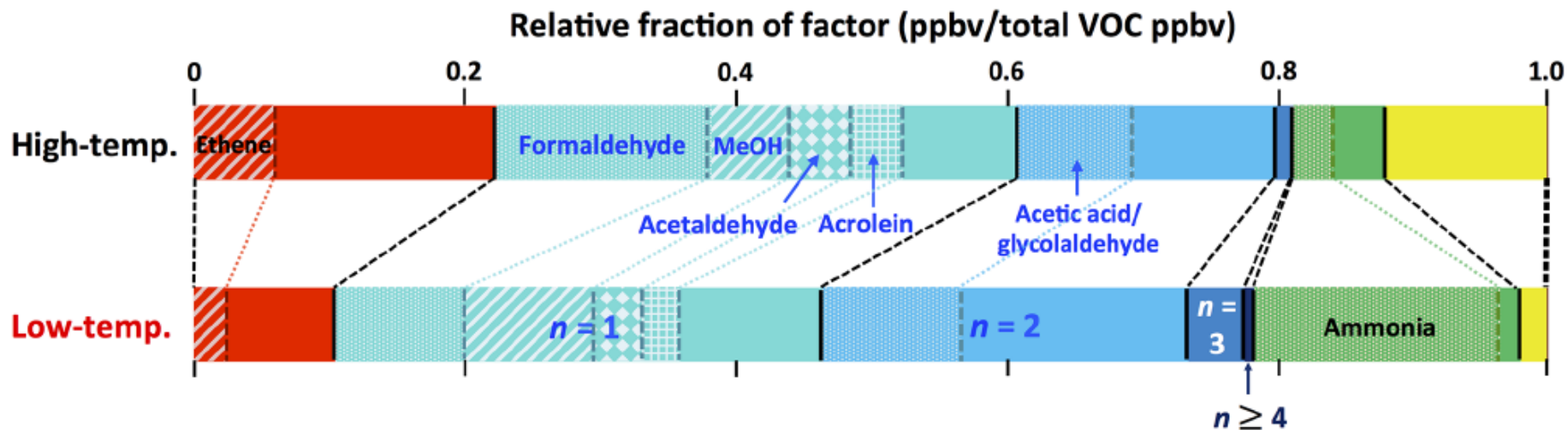


Biomass Burning Sources of NH₃



Poster by Viney Aneja

Biomass Burning Sources of NH₃



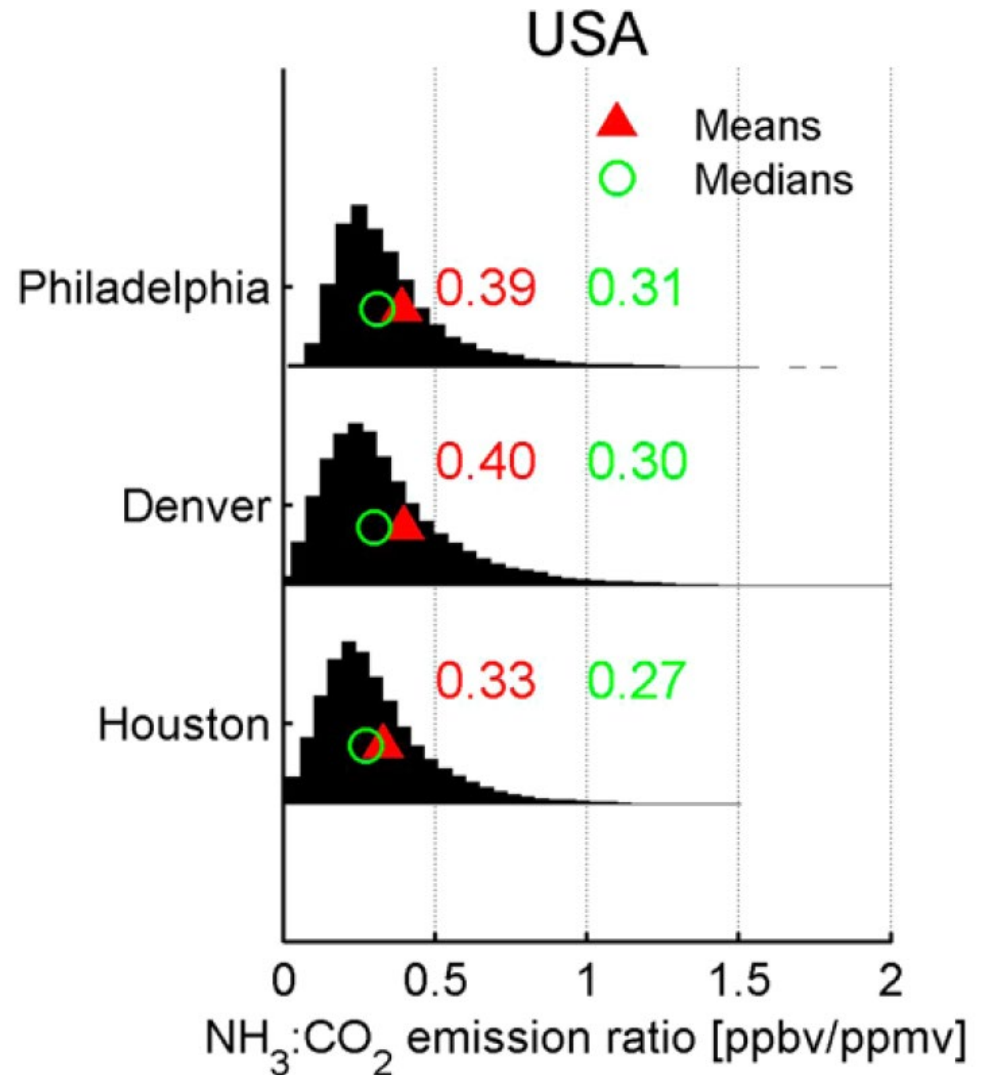
	High-T	Low-T		High-T	Low-T
Hydrocarbons	0.223	0.105	N-containing	0.067	0.196
Oxygenates (number of oxygen : n)			N and O-containing	0.119	0.019
$n = 1$	0.386	0.359	S-containing	0.001	0.001
$n = 2$	0.190	0.270	Others	0.000	0.000
$n = 3$	0.014	0.041			
$n \geq 4$	0.001	0.008			

Mobile Sources of NH₃

Can be diagnosed through emission ratios with CO, CO₂, NO_x

Gasoline vs diesel?

Differences in emission ratios during winter operating conditions?



Key Questions

- How does the sensitivity of $\text{PM}_{2.5}$ to NH_3 vary
 - spatially across the western U.S. in winter?
 - through time over the course of an extreme episode?

Requires: extensive (ground-based, aircraft, satellite), high-time resolution (hourly or better) measurements of NH_3 and p- NH_4^+ (plus HNO_3 and other $\text{PM}_{2.5}$ chemistry)

- What are the emissions of NH_3 for each sector and how do they vary seasonally?

Requires: extensive (ground-based, aircraft, satellite), high-time resolution (hourly or better) measurements of NH_3 and p- NH_4^+ (plus co-emitted species CO , CO_2 , CH_4 ...)

NH₃ concentration at SLC (AMoN)

